

中国科学院大速化李衡理研究的

DALIAN INSTITUTE OF CHEMICAL PHYSICS, CHINESE ACADEMY OF SCIENCES

# Recent R&D Activities on Clean Coal Technology in DICP

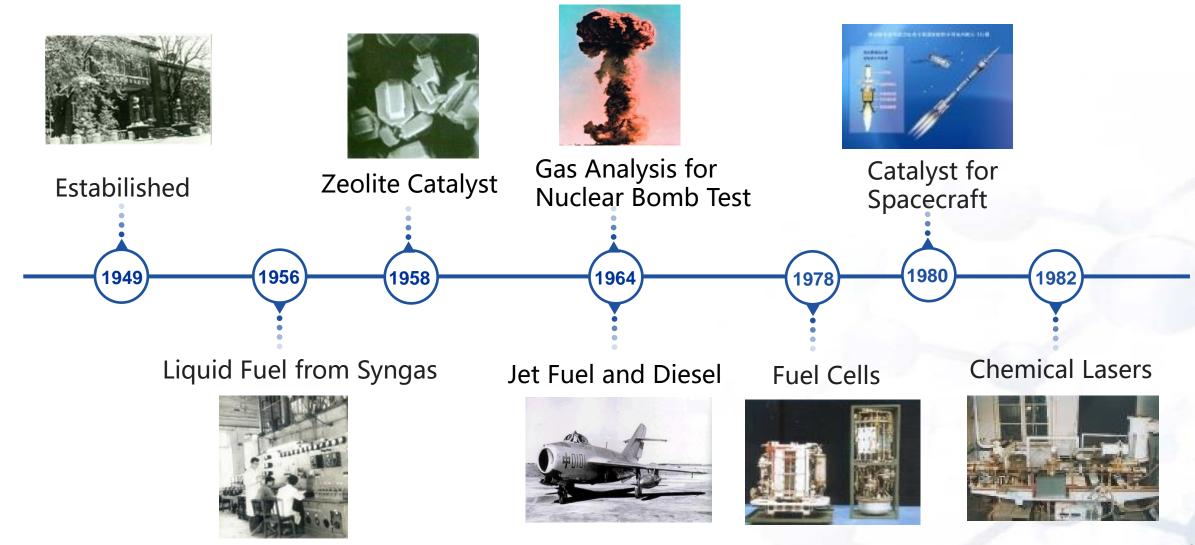
Rui CAI

Nov. 30<sup>th</sup>, 2017

- Brief Introduction of DICP
- Recent R&D Activities on Clean Coal Technology in DICP

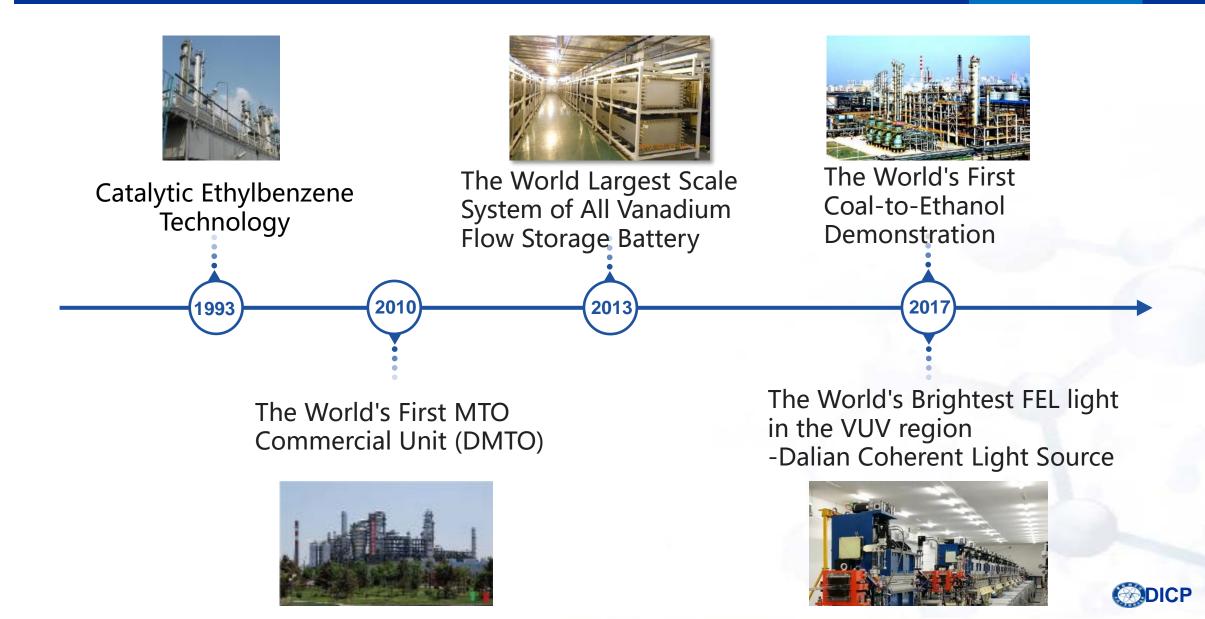


# **History - An Institute of Chinese Firsts**

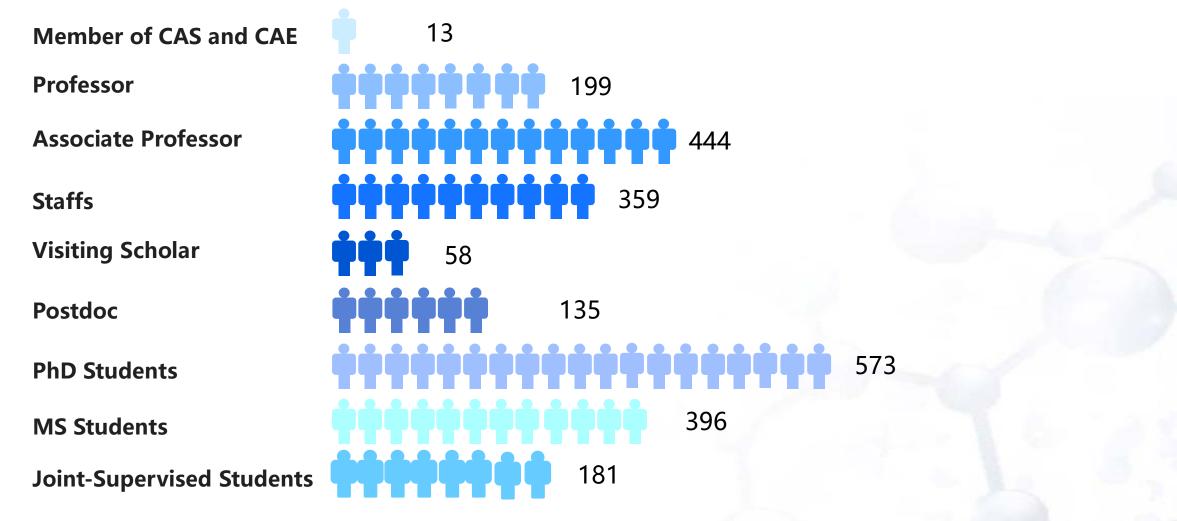




## **History - An Institute of Chinese Firsts**



## **Human Resources**



~2300 working and studying in DICP



## **Members of CAS and CAE**



Dayu ZHANG (1906-1989)

**Qingshi ZHU** 

(1946-)



**Xiexian GUO** (1925 - 1998)

Liwu LIN

(1960-)



**Peizhang LU** 



**Guohe SHA** (1934-)









Shengli YANG

(1941-)

**Cunhan ZHANG** 

(1928-)





**Baolian YI** 

**Guozhong HE** 

(1933-)





(1934-)



**Nanquan LOU** (1922-2008)



Yukui ZHANG (1942-)





(1964-)





## **General Information of DICP (2012-2016)**





**10** International Research Centers



4000 Publications



**1100** Graduate Students



5000 Million Research Funds



4000 Filed Patents



8 Research Laboratories



**29** Spin-off Companies





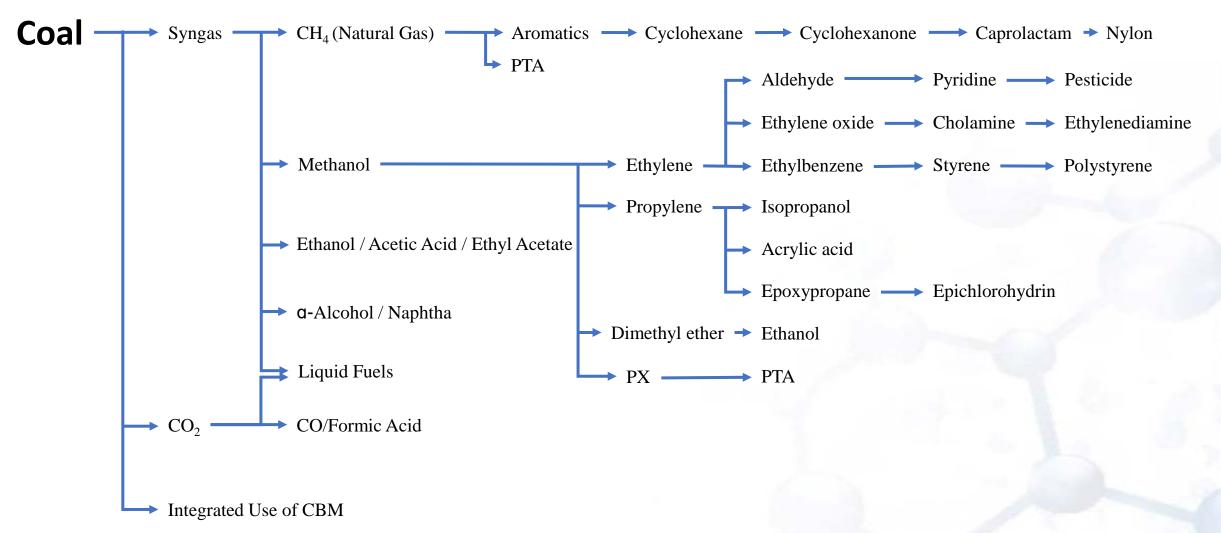


DICP

- Brief Introduction of DICP
- Recent R&D Activities on Clean Coal Technology in DICP

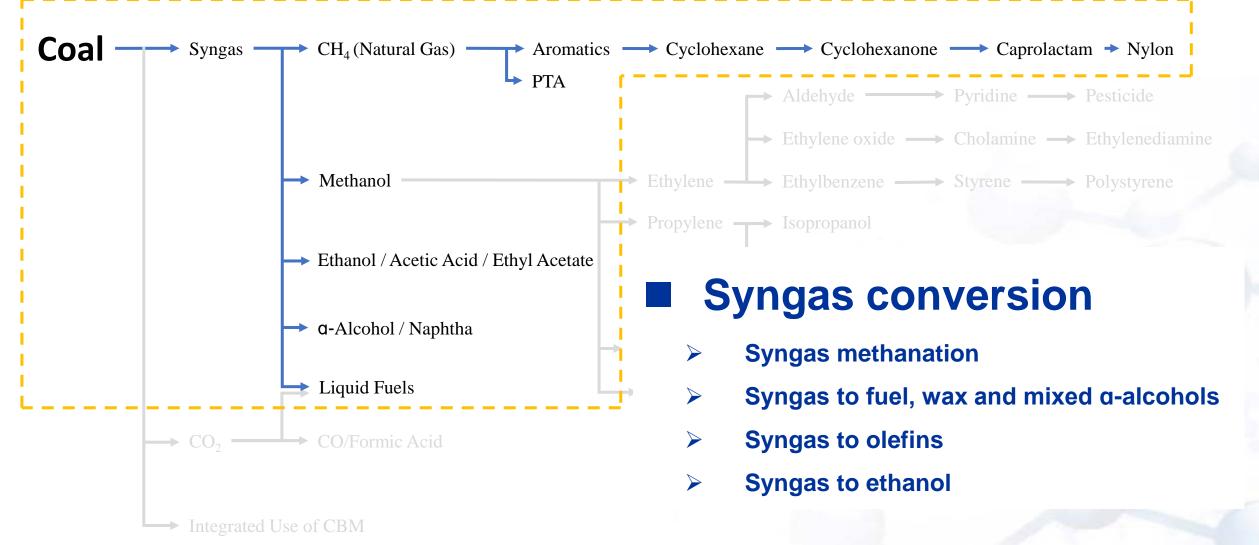


## **DICP Major Activities in Clean Coal Researches**



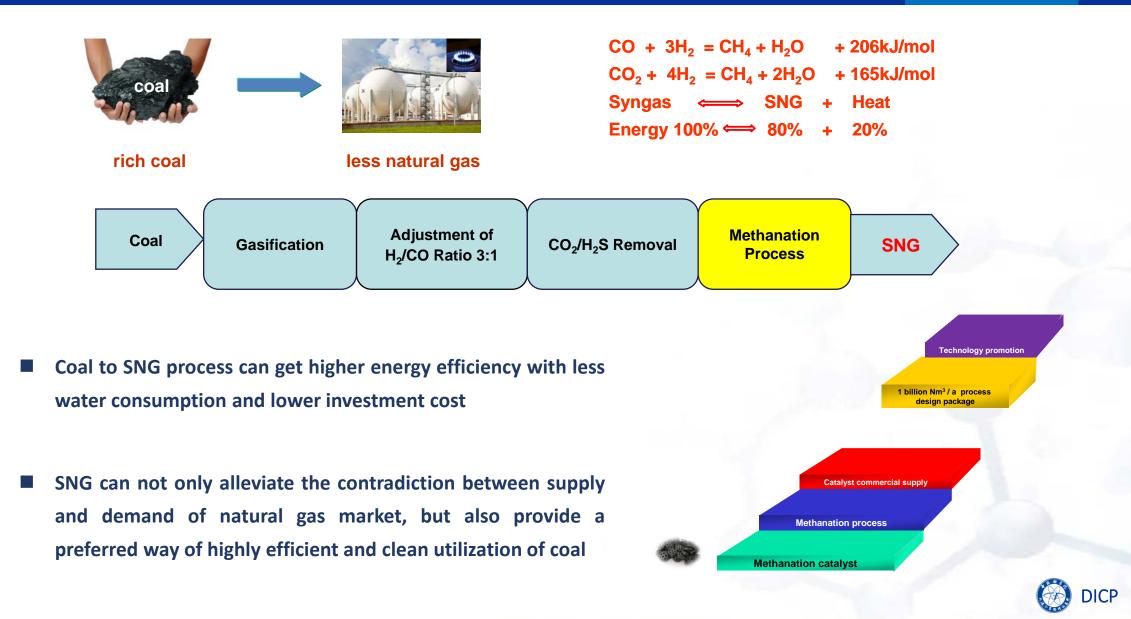


## **DICP Major Activities in Clean Coal Researches**



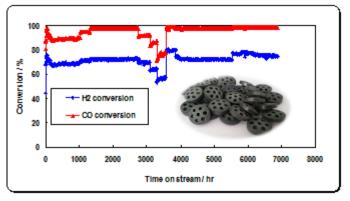


## **SNG Production from Syngas Methanation**

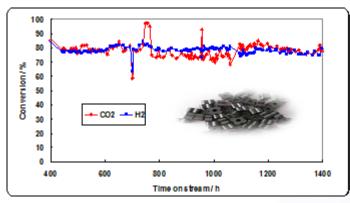


## **Breakthrough of Key Technologies**

• DICP Methanation Catalyst (13 Pieces of Chinese Patent Application)

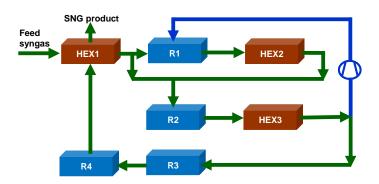


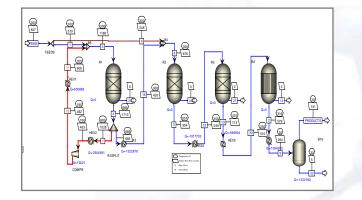
The high temperature methanation catalyst was developed based on the novel synthesis technic and the materials with good hydrothermal stability which effectively prevented sintering of the nickel crystal. Catalyst had been tested with a total of 7,500 operating hours in the temperature range of 600-700 °C.



The low temperature methanation catalyst has good activity and stability which ensure maximum conversion of CO and CO<sub>2</sub>. The catalyst had been tested with a total of 1,500 operating hours in the temperature range of 250-450 °C.

#### • DICP Methanation Process (4 Pieces of Chinese Patent Application)





The multi stage methanation process based on DICP's novel methanation catalyst and its methanation reaction kinetics, addressed the essential question of reaction heat recovery and was also economic competitiveness for investment and operating costs.

12

DICP

## **Providing A Total Solution**

• Demonstration Project



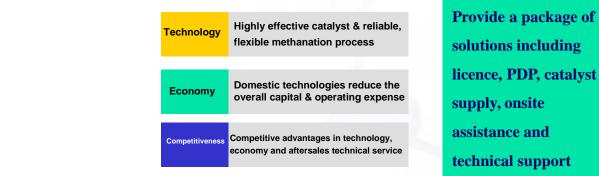
Demonstrate device in Yima, Henan Province, with a capacity of 200 Nm<sup>3</sup>/h syngas



Demonstrate device in Guanghui Ltd., Xinjiang Region, with a capacity of 6000 Nm<sup>3</sup>/d SNG

- > DICP's proprietary methanation catalyst as well as process had been validated and demonstrated in two pilot scale plants under realistic industrial conditions.
- > The total running time had been up to more than 4,400 hours and SNG product meet with the requirements of the national pipeline natural gas.
- > Based on the results, 1 billion Nm<sup>3</sup>/a SNG process design package had been compiled and a preliminary review had been completed.
- Catalyst Production

A catalyst manufacture plant with the annual output of 50 tons was built in Dalian and thus DICP has the ability to supply commercial methanation catalyst.



## **Multi-techniques for Ethanol Production from Coal**



1 kt/a pilot plant test of ethanol synthesis from syngas have conducted in 2016



30 kt/a demon for hydrogenation of acetic acid to ethanol have done in 2016



**300 kt/a ethanol and iso-propanol production from acetic acid/propene as feed-stocks have done in 2015** 



Pilot plant test for hetero-oxo-synthesis of methanol and syngas and its hydrogenation to ethanol

DICP

## Syngas to Fuel, Wax and Mixed a-alcohols

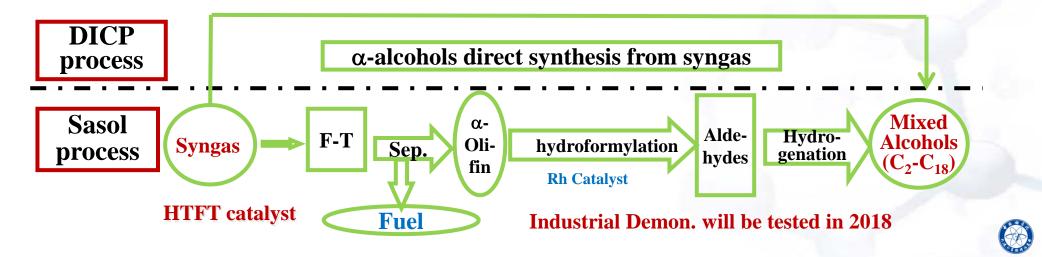
**3kt/a** Demon test have done using a Co/SiO<sub>2</sub> catalyst and a fixed bed reactor in 2007



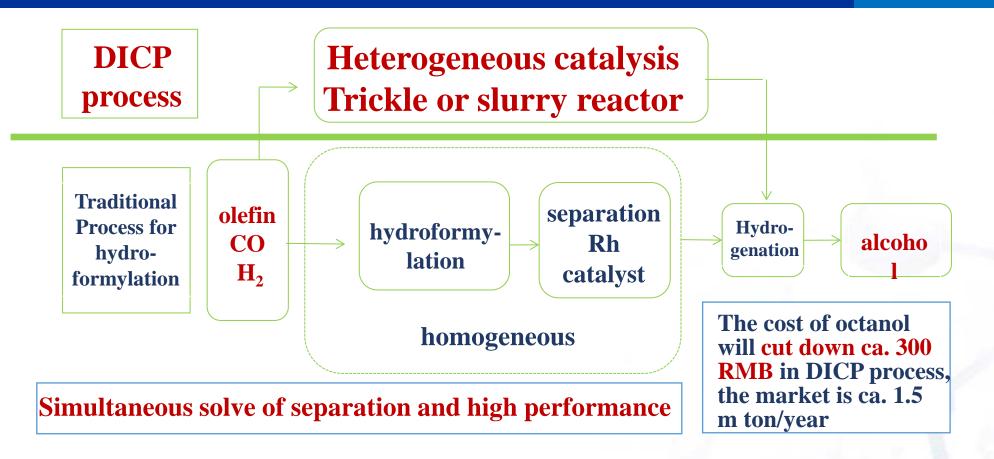
200kt/a 3 industrial facilities are being built in China using Co/SiO<sub>2</sub> catalyst and fixed-bed reactor



150 kt/a demonstration have been succeed to produce selectively naphtha and diesel using Co/AC catalyst and slurry reactor in 2015-2017.

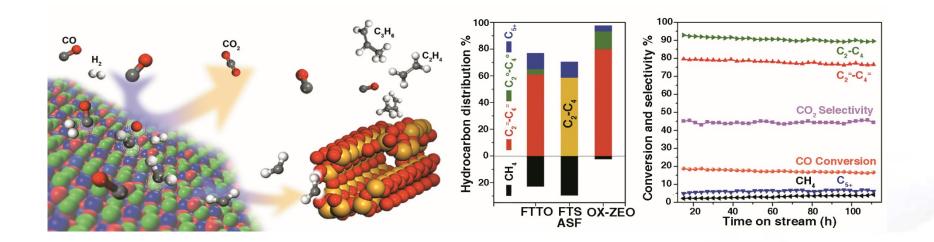


### **Heterogeneous Hydroformylation of Olefins to High Alcohols**



- Fundamental understanding at molecular level to setup and develop theory for single sites catalysis.
- Novel catalysts with high performance will be developed, the pilot tests for C3=/C4= hydroformylation reactions will be conducted.

## **Converting Syngas Directly to Light Olefins**

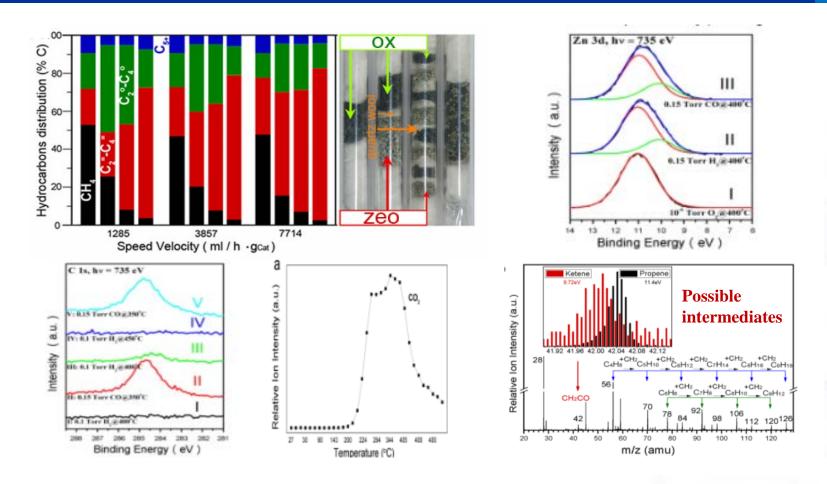


Two different types of active sites for CO activation and C-C coupling. Leading to selectivity beyond ASF distribution limit.

Remove oxygen with CO, possibly circumventing energy-intensive water-gasshift process.

> Jiao, Li, Pan, Bao et al., **Science** 351(2016)1065. **Perspective,** De Jong, **Science** 351 (2016) 1030.

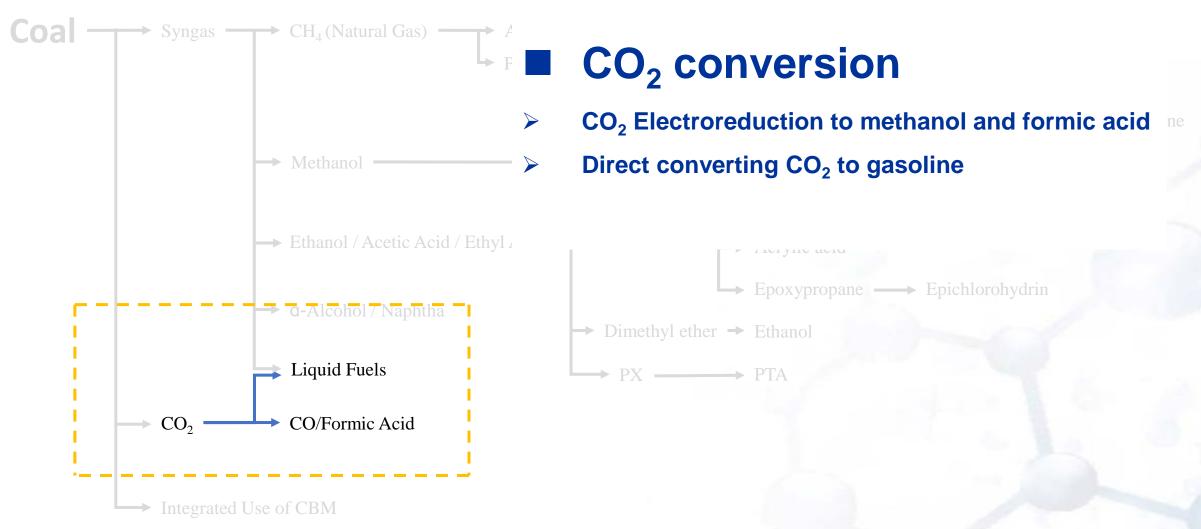
## **Reaction Mechanism**



The role of oxides and zeolites, activation of CO, and possible reaction intermediate were studied by in situ-XPS, synchrotron-based vacuum ultraviolet photoionization mass spectrometry (SVUV-PIMS), etc.

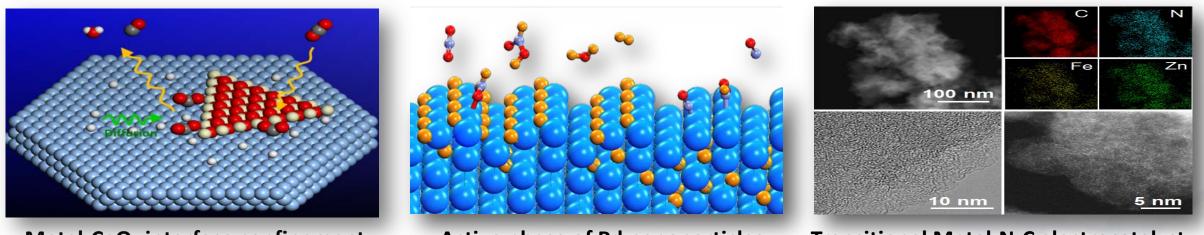


## **DICP Major Activities in Clean Coal Researches**





## **CO<sub>2</sub> Electroreduction to Formic Acid and CO**



Metal-CeO<sub>2</sub> interface confinement

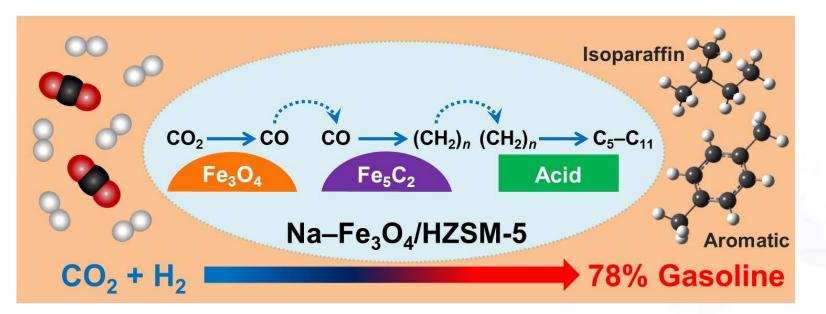
Active phase of Pd nanoparticles

Transitional Metal-N-C electrocatalyst

- □ Liquid Fuels and chemicals production from CO<sub>2</sub> and H<sub>2</sub>O with electricity from renewable energy or abundant nuclear energy
- □ Simplified process, without H<sub>2</sub>, reaction temp: r.t. to 800 °C
- Electricity storage and carbon recycling simultaneously

*J. Am. Chem. Soc.*, 2015, 137, 4288; *Nano Energy*, 2016, 27, 35; *Nano Research*, 2017, 10, 2181; *J. Am. Chem. Soc.*, 2017, 139, 5652; *Nano Energy*, 2017, 38, 281; *Chemical Science*, 2017, 8, 2569.

## **Direct Converting CO<sub>2</sub> to Gasoline**



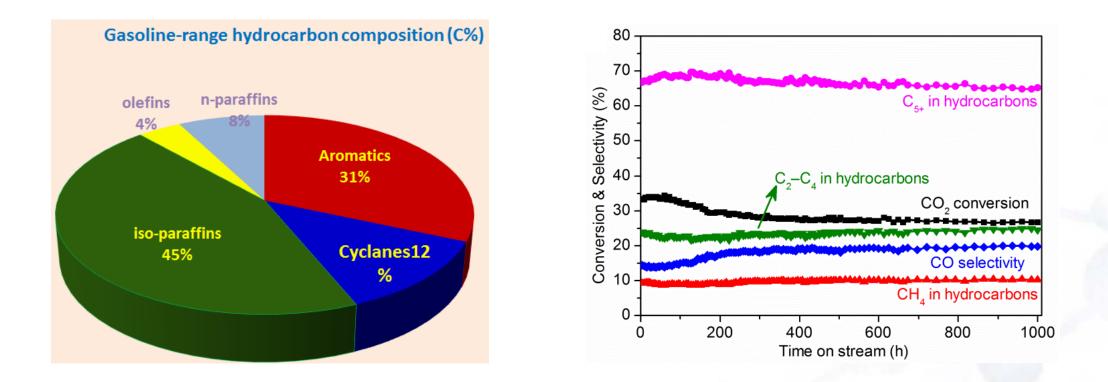
Direct hydrogenation of  $CO_2$  into liquid fuels can mitigate  $CO_2$  emissions and reduce the rapid depletion of fossil fuels. Here we show a multifunctional catalyst that converts  $CO_2$  to gasoline with high selectivity due to synergistic catalysis of active sites.

#### "This work can be considered as a breakthrough in CO<sub>2</sub> catalysis" (referee)

J. Wei, Q. Ge,\* R. Yao, Z. Wen, C. Fang, L. Guo, H. Xu, J. Sun\* Directly Converting CO<sub>2</sub> into a Gasoline Fuel. *Nat. Commun.* 2017, 8, 15174 *Selected as a Research Highlight in Nature* 2017, 545, 7653



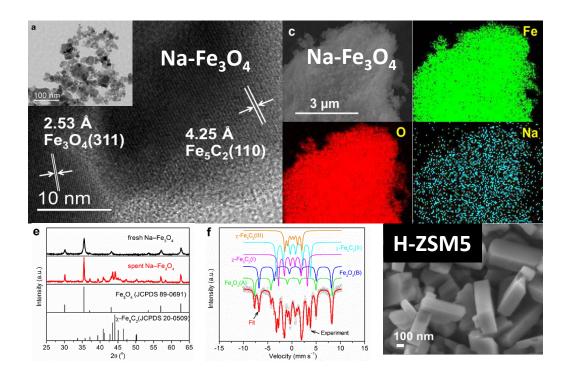
## **Direct Converting CO<sub>2</sub> to Gasoline**



- Product: Conformed to China V-5 Gasoline standard
- **Catalyst: Well stability after 1000 h running**



## **Direct Converting CO<sub>2</sub> to Gasoline**



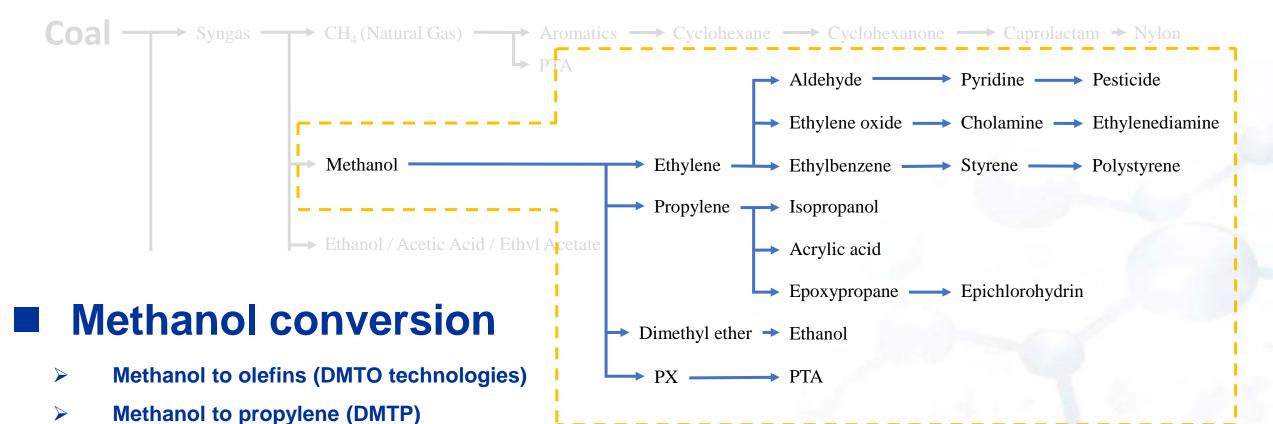
Catalyst	т	Р	X <sub>CO2</sub>	со	Hydrocarbon /%		
	l∘C	/MPa	/%	/%	CH <sub>4</sub>	C <sub>2-4</sub>	<b>C</b> <sub>5+</sub>
FeZnZr/HZSM-5	360	5	19	57	2	46	52
FeZnZr@HZSM-5	340	5	14	41	2	64	34
Fe-Ce/KY	300	1	20	35	9	41	49
Fe/RbY	300	1	17	32	9	36	54
Our work	320	3	22	20	4	16	80
FeCuNa/HZSM-5	250	2	12	20	29	42	29
CuZnAl/HB	300	1	27	53	2	93	5
FeZnZr/HY	340	5	22	50	2	84	14
CuZnZr-Al/Pdβ	260	2	25	47	1	77	21

Confirmation of active sites: Na-Fe<sub>3</sub>O<sub>4</sub>/Fe<sub>5</sub>C<sub>2</sub>/zeolite

Gasoline sel. reaches the highest value among reports



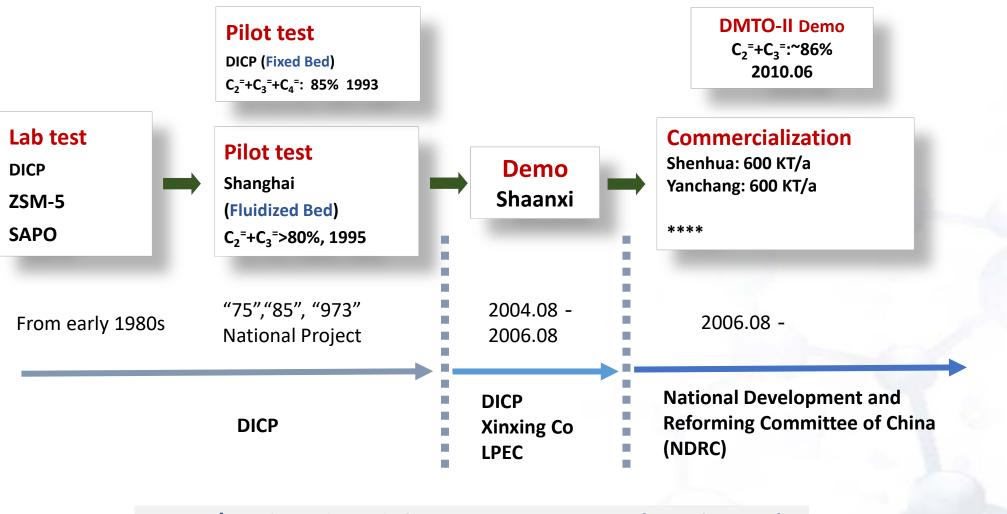
## **DICP Major Activities in Clean Coal Researches**



- Co-production of p-X and olefin by methanol and toluene
- Ethanol from DME carbonation and hydrogenation
- Methanol-coupled-naphtha to light olefins



## **Roadmap for DMTO Process Development**



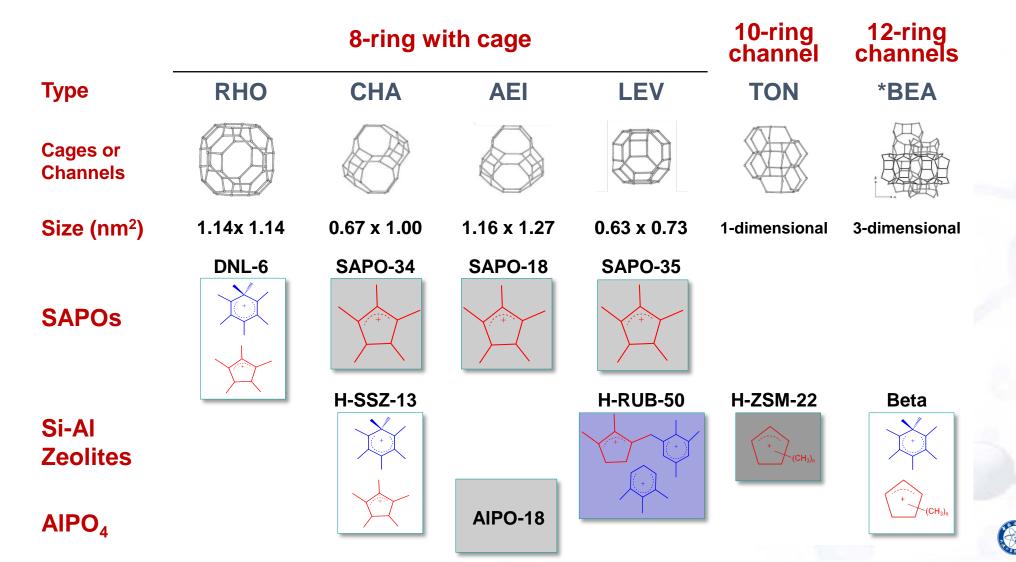
DME/Methanol to olefins = DMTO

(DICP's MTO)



### **Progress on Reaction Mechanism**

#### **Carbenium ions in different cages and channels have been observed**

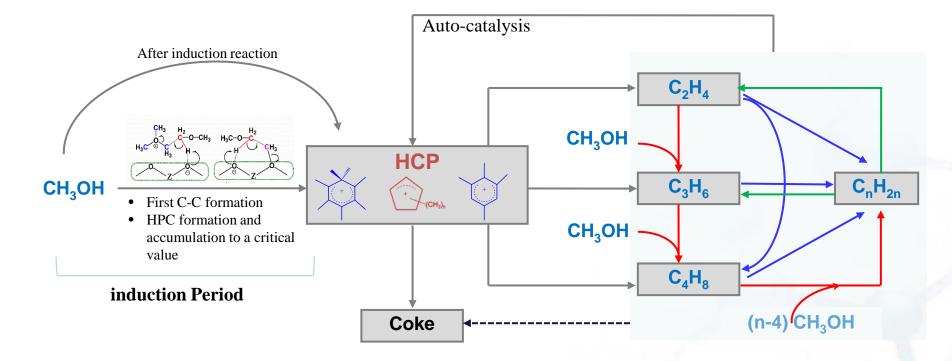


26

DICP

## **Summary on Mechanism**

#### The mechanism picture of the reaction



- Methylation reaction could not be avoid
- Cavity (size and environment) controls the selectivity
- Acidity changes the balance between the mechanisms



## **Scale-up of DMTO Reactor**



**MeOH Feed Cat Inventory** 0.01 kg Ugas **Fluidization** 

**Micro Scale** ~1.2 kg/d ~2-- 25 cm/s

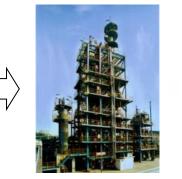
**Bubbling bed** 



**Pilot Scale** ~120 kg/d 1 kg

~5-- 25 cm/s

**Bubbling bed** 



**Demo Scale** 50,000 kg/d 300 kg ~ 1-2 m/s

**Turbulent bed** 



**Commercial Scale** 5500,000 kg/d 45,000 kg ~1-2 m/s **Turbulent bed** 



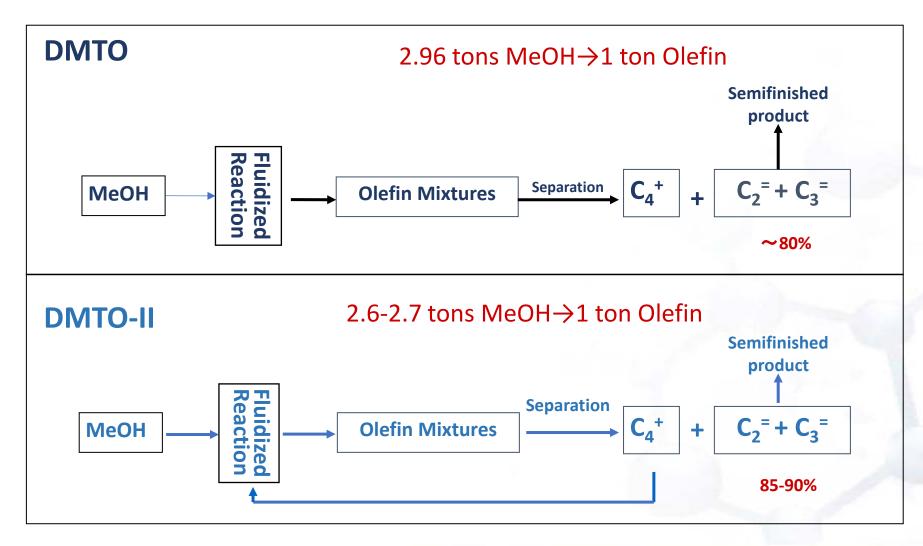
## **DMTO Demonstration Results: 72 Hours Calibration**

Feedstock	CH₃OH	
Scale	50t/d	
Reactor type	Fluid bed	
Single pass conversion %	>99%	
Yield of ethylene and propylene, wt%	33.73	
Selectivity of ethylene and propylene, wt%	>79.1	
Feedstock consumption for each ton of ethylene and propylene, t/t	2.96	
Catalyst	D803C-II-01	



## New Generation of DMTO Technology (DMTO-II)

#### **Technical Principle**



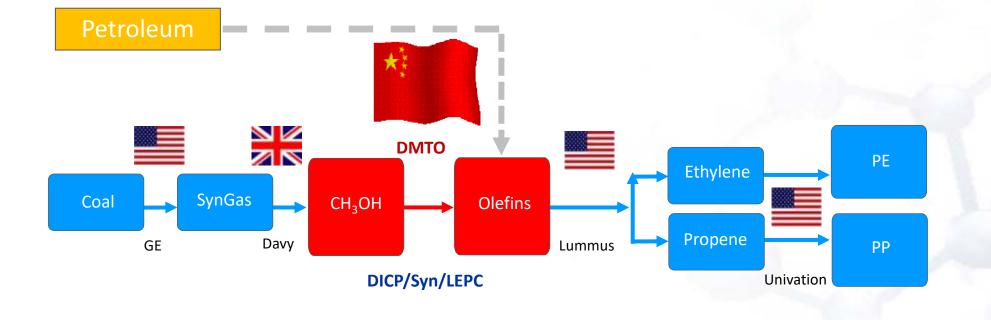
30

DICP

## **Commercialization of DMTO**

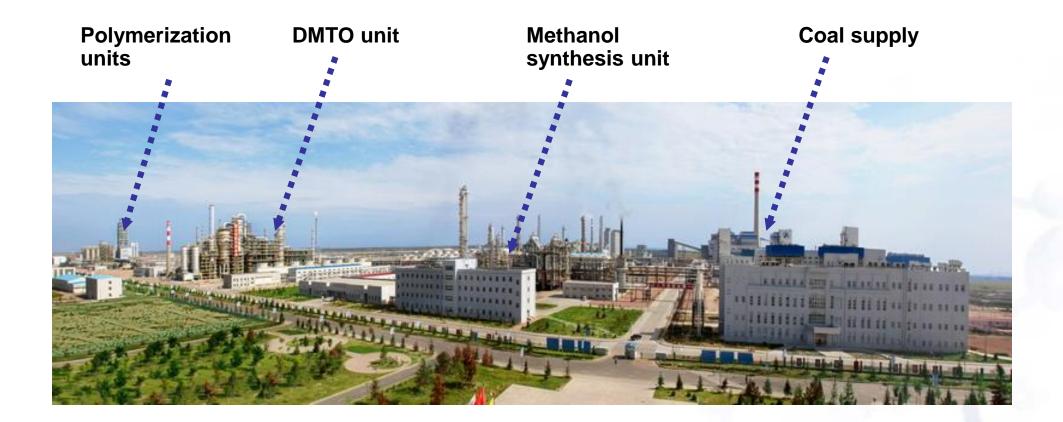
#### Baotou Coal-to-Olefin Project of China Shenhua Group

- 1,800 KTA Methanol  $\rightarrow$  600 KTA polyolefin
- Approved by NDRC (Dec, 2006)
- Construction was finished on May 31, 2010
- First coal to olefin plant in the world





#### World First Coal to Olefin Plant in Baotou of China Shenhua Group

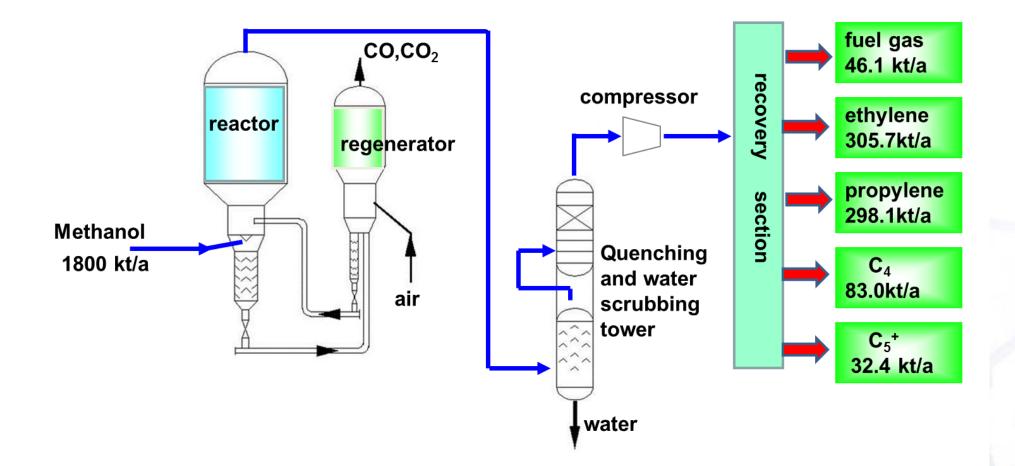




#### **DMTO Unit of Coal to Olefin Project in Baotou of China Shenhua Group**



## **Typical Products in DMTO Plant (1.8 Million Tons MeOH)**

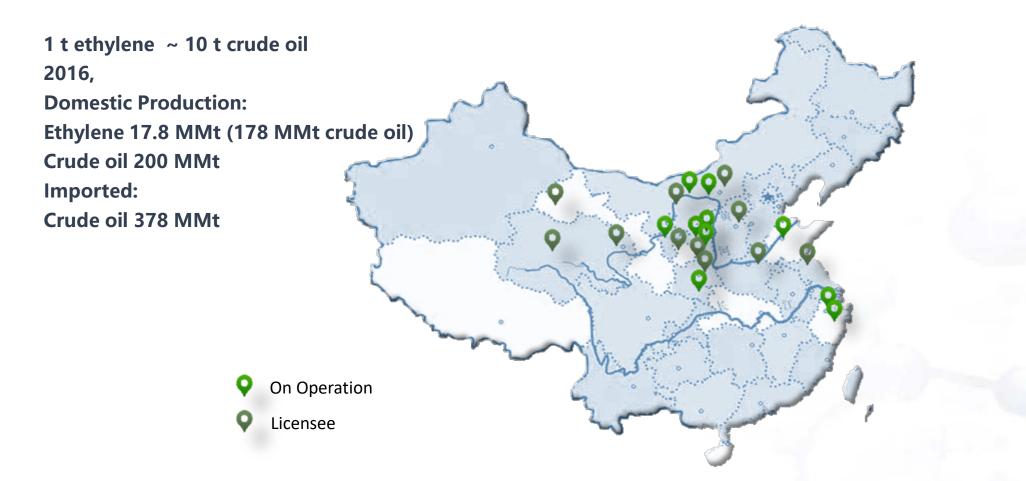


Methanol consumption for 1 ton olefin is 2.96 ton

Ethylene and propylene selectivity ~80%



# **DMTO Plants in China**



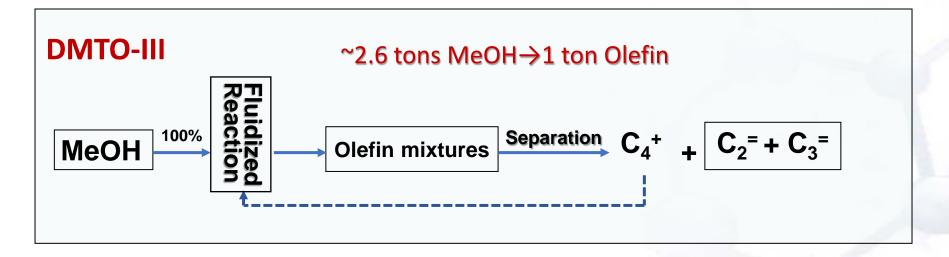
So far, DMTO technology has been licensed in 24 commercial units (14 MMt/a ) in the domestic market. 12 commercial installations (7MMt/a ) have commissioned in just five years



## **DMTO-III Technology**

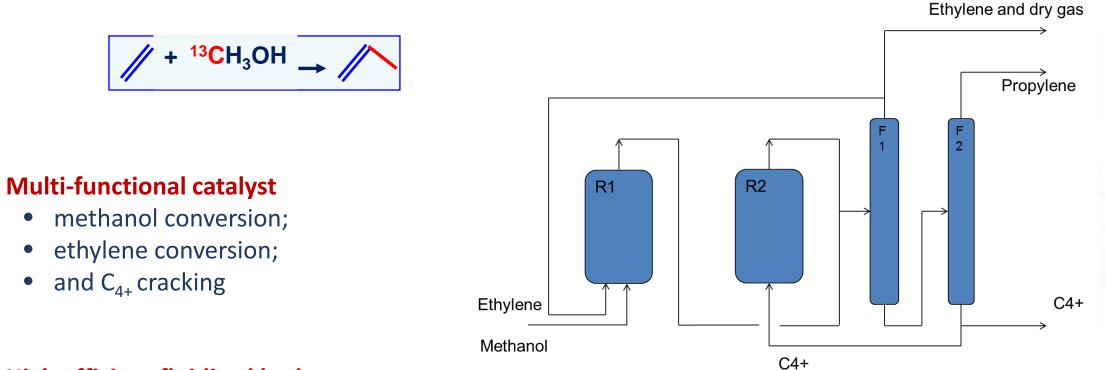
#### **Technical features**

- New generation of DMTO catalyst
- Higher operation pressure
- One unit:  $3 Mt/a \longrightarrow 1.15 Mt/a$  light olefin
- Ethylene + Propylene Selectivity: ~90%





## **DMTP technology**

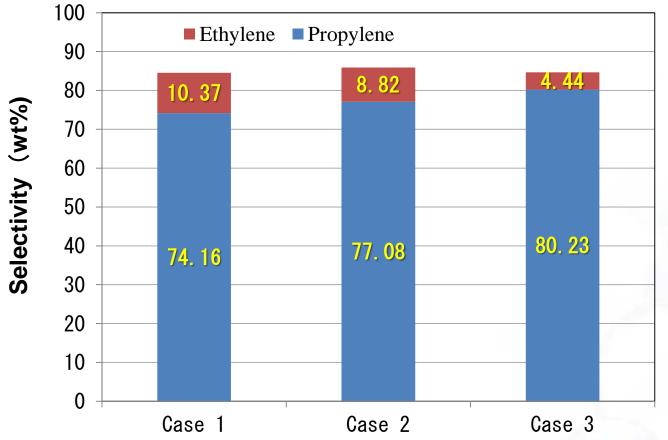


37

DICP

- **High efficient fluidized bed reactors** 
  - methanol and ethylene conversion in one reactor
  - adjustable propylene yield

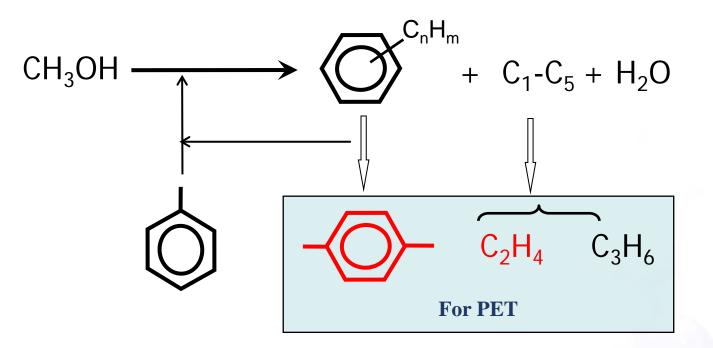
## Simulation results based on pilot tests (1t/d)



(MeOH Conversion=100%)



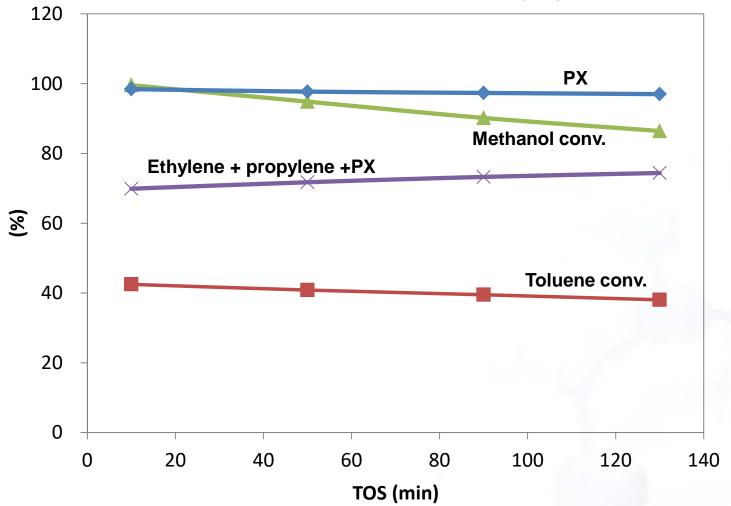
### **Co-production of p-X and Olefin by Methanol and Toluene**



- Why co-production?
- Fluidized bed reactors



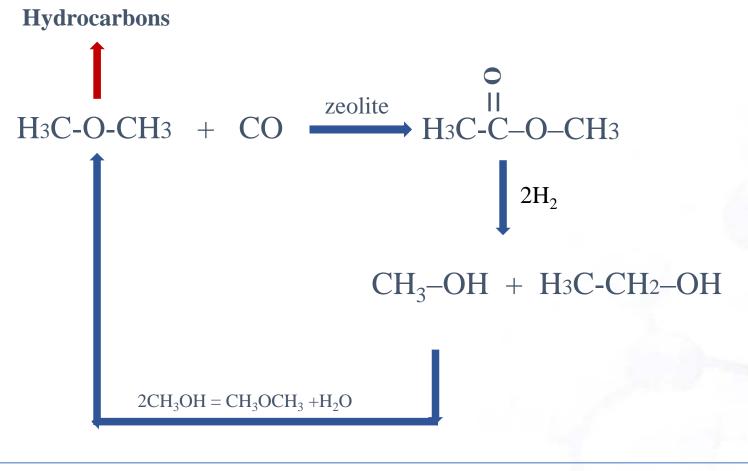
## **Typical Results**



M/T= 10 (mol), MeOH WHSV =  $1.6 h^{-1}$ 



## **Ethanol from DME Carbonation and Hydrogenation**



Total:

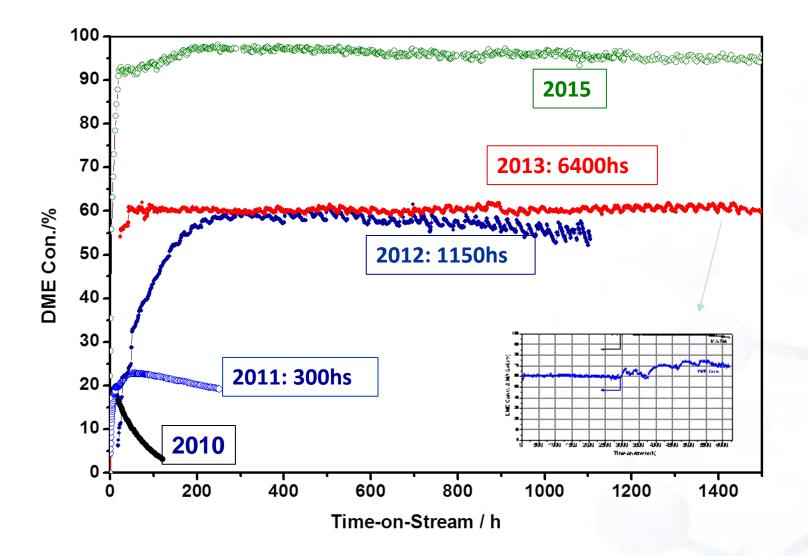
 $CH_3OCH_3 + 2CO + 4H_2 = 2CH_3CH_2OH + H_2O$ 

Total from syngas :

 $2CO + 4H_2 = CH_3CH_2OH + H_2O$ 



## **Improvement of Catalyst Life**





## **Ethanol from DME Carbonation and Hydrogenation**



The world's first Coal-to-Ethanol (methanol to ethanol) Demonstration with 100,000 metric tons of pure ethanol per year in January 2017





- Systematic research and development have been processed on clean coal utilization in DICP, providing technical support to harmonious development of coal chemical industry and petrochemical industry.
- Several technologies have been utilized in industrial implementation, such as DMTO, ethanol technology, etc.
- The industrialized projects stood the tests of fluctuations in oil prices and develop rapidly, which show excellent market competitiveness.
- Although great achievements have been obtained in Coal Chemical Industry in China, there still has tremendous development space in the future.
- Technical innovation is always the most important for the world's green growth and sustainable development.









